UniduringHED PRELIMINARY DATA

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Effects of ecidity on call division 4. 27 W68 85 10 1

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It has been reported that carbon dismide may act as a powerful inhibitor of call division (2). If synchronized 7-hour old cells of Chierelle 7-11-05 are resuspended in distilled water, the division readily proceeds in dark in exmospheric air but not in air enriched with carbon diomide. The inhibitory action of carbon dioxide is effected by other ions. Thus, in bicerbonete buffers, a molar some contration of bicarbonete equal to or higher than that of carmon dioxide in the suspending fluid completely suppresses the edverse ection of carbon dismide on cell division. The belancing action of bicarbonate extends up to I x 10-2 M CO2. At higher concentrations of carbon dioxide, bicarbonate is unable to suppress CO2 inhibition of call division. Other acids act similarly on cell division. So far the effects of hydrochloric, nitric, sulfuric, phosphoric, ecatic, glycolic, and eliric acids have been studied.

In these experiments, synchronized cells of the green, highcorperature alga, Chiorella 7-11-05 (1), were grown under optimal

conditions. After 7 hours in light, cells were contrifugué and transferred into a suspending fluid containing a certain concentration of
acid. Such buffered and unbuffered solutions were used. Cell suspensions were then placed in derivass and supplied through bubbling tubes
either with examphoris air or with gas sixtures containing different
percentages of earten disside in example eir.

coff division was massured by counting undivided cells. This manher was then subtracted from the original number of mother cells in a given volume of cell suspension and the difference was empressed in per cent of the original number of cells. The results of three series of experiments, based on observations on cells kept under given conditions in derivous for 16 hours, are presented in table !.

to one series of experiments, cells were suspended in distilled were or in 1 x 10⁻⁵ H and 1 x 10⁻⁶ H sulfuric acid. The suspending fluid was unbuffered and the cell suspension was supplied during division time in derimass with atmospheric eir. Cell division proceeded setisfactorily is distilled water and in 1 x 10⁻⁶ H H₂SO₆, but not in 1 x 10⁻⁶ H H₂SO₆, which proved to be inhibitory to cell divisions.

In the second series of experiments, the suspending fluid was buffered with 1 x 10⁻¹² N Tris (hidroxymethyl) aminomethane buffer and supplied during division time, as in series one, with atmospheric air. Here the division took place in a wide range of concentrations of sulfuric acid up to 1 x 10⁻¹² N M₂SO₄. The adverse effect of sulfuric acid was conceiled by the buffer. However, an increase in M₂SO₄ consentration by 1 x 10⁻¹⁶ N tipped the belance between H₂SO₄ and "Tris" and sewed inhibition of cell division.

Table !

In the third series of experiments, calls were supplied in dark with I per cent CD_2 in air mixture. Calls suspended in distilled worse did not divide in the absence of buffer. Thus, one per cast CD_2 (2.4 x 10^{-4} N) proved to be inhibitory to call division. The particular age of cells divided in I x 10^{-2} N and in 1.01 x 10^{-2} N K_2SD_4 was former than to atmospheric air (Series II), indicating that the effect of carbon disting was added to that of sulfuric acid.

For a given batch of cells, the inhibitory acid concentration has a precisely determinable value. However, in a series of experiments is became evident that the acidic action on tall division is affected by other factors which are under study in our laboratory.

REFERENCES

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- 1. This work was supported by funds from the Marizonsi reconstitus and Space Administration.
- 2. Scientific Article A974, Contribution No. 3356 of the University of Maryland Agriculture: Experiment Station.

Table 1. Call division in synchronized 7-water calls of planes as some efter 16 hours in derivess at 35° .

| | Entirending | | | Por seal |
|--------|-------------------------|------------------|-----------------|---|
| Series | fluid | B uffer | | |
| 2 | KgO | No | Aumospheric bir | |
| | 1 x 10-5% H2504 | ₹ \$ | < & | 23 |
| | 1 x 10 ⁻⁴ *4 | \$? | \$11 | |
| | n _z o | l a 10°4 Tris | 2 0 | i de la companya de |
| | 1 x 10-4 11250 h | ₹ | € § | * |
| | 1.01 x 10-2 :: | *: | f a | \$ (24) |
| | ×,0 | No. | | |
| | H ₽ | 1 x 10° 2× 10 (x | . # | <u>Z</u> - |
| | 1 x 10 ** ×, \$0, | . 3 | | |
| | 1.31 × 10 *** | 4.6 | # T. | <u>~</u> |